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Amidon et al. published the importance of **Biopharmaceutical Clasification System (BCS)** in 1995, which is an important characteristic of the preparation's bioavailability.

According to their suggestion APIs can be categorized into four groups regarding the gastrointestinal absorption by their solubility and permeability characters.

APIs can be categorized into four groups according to their solubility and permeability:



streght across the physilogical pH range (1.2; 4.5; 6.8)



APIs with high solubility and high permeability

API absorbs well (though the metabolism can decrease the bioavailability), the rate controlling step will be the dissolution.

 $k_{diss} << k_{abs}$

APIs with high solubility and high permeability





acetaminophen, paracetamol

propranolol



verapamil

imipramine

APIs with high solubility and high permeability



ephedrine

APIs with low solubility and high permeability

API's absorption is restricted by the dissolution's speed and amount.

Because of the low solubility the absorption occurs at the GI tract's longer regions, the dissolution lasts for a longer period of time and it influences the concentration at the place of absoprtion.

$$k_{diss} \leq k_{abs}$$

APIs with low solubility and high permeability



APIs with low solubility and high permeability



APIs with high solubility and low permeability

The transfer speed through the membranes is very important in this case.

Long contact time with the mucus is required to increase the absorption (mucoadhesive systems).

 $k_{diss} >> k_{abs}$

APIs with high solubility and low permeability



atropine

ranitidine

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APIs with high solubility and low permeability



APIs with high solubility and low permeability



BCS Class IV

APIs with low solubility and low permeability

Absoprtion of APIs belongign into this group is low, thus peroral administration is doubtful.

Technologies increasing the solubility should be applied: molecular dispersion, solubilisation.

BCS Class IV

APIs with low solubility and low permeability







BCS Class IV

APIs with low solubility and low permeability



hidrochlorotiazide

BCS Class Membership of Selected Model Drugs



Volume (ml) of required to dissolve the highest dose strength at the lowest solubility in pH 1-8 range (Note: 1 ml set as low value)

Three type of rates can be introduced according to the BCS that help researchers to develop preparations:

- absorption rate (A_n)
- dose rate (D_o)
- dissolution rate (D_n)

Absorption rate (A_n) – ration of residence time in the small intestine and the time needed for absorption:

$$A_n = \frac{t_{res}}{t_{abs}}$$

 t_{res} residence time t_{abs} time needed for absorption **Dose ratio** (D_o) – can be calculated by the solubility, dose and tha amount (volume) of water used for the administration:

$$D_o = \frac{D}{V_o C_s}$$

- D dose,
- V_o volume of water used at the administration,
- C_s solubility.

Dissolution ratio (D_n) – is the ratio of API's residence time in the GI tract and the time of dissolution:



 t_{res} residence time, t_{diss} dissolution time,

Literature shows that there is an increasing number of poorly soluble compounds in the drug discovery pipeline – including 70 per cent BCS Class II compounds – although only about 30 per cent are BCS Class II in the existing top 200 market



Sources of pharmaceutical technology to increase the bioavailability of drug delivery systems accourding to the BCS



Many THANKS for Your Attention Dziękuję Ďakujem dhanya-waad Дякую bedankt ありがとう go raibh maith agat tesekkürle Спасибо Thank you آنگر آ 前前 Merci köszi tack så mycket Thank you faleminderit Shukriyâ Danke hvala díky kiitos takk Obrigada Mulţumesc nandri Ευχαριστώ Grazie anugurihiitosumi dhanya-waad Muchas gracias köszönöm Terima Kasih ačiû 너를 감사하십시요 mange tak aitäh děkuji vam salamat